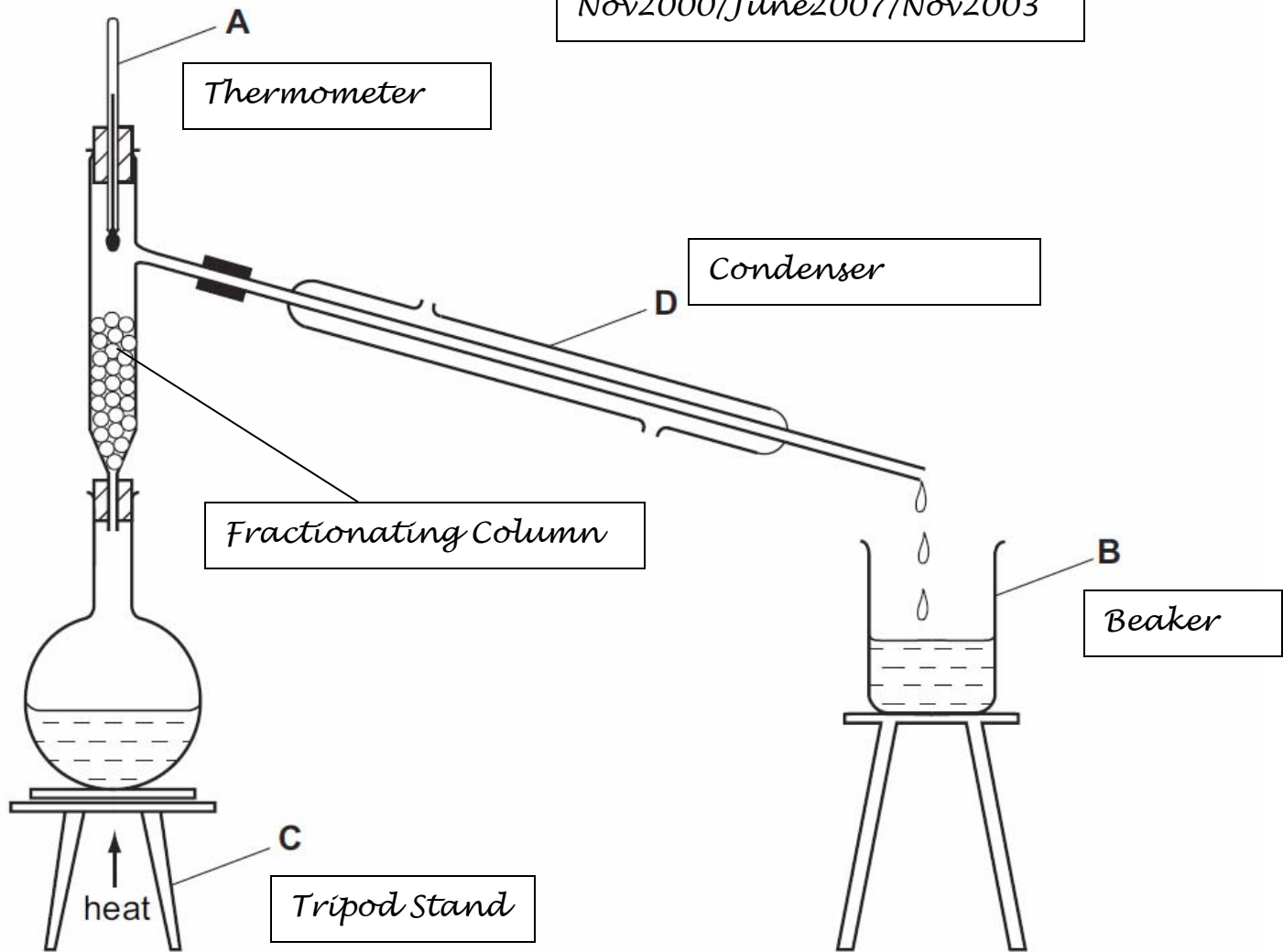
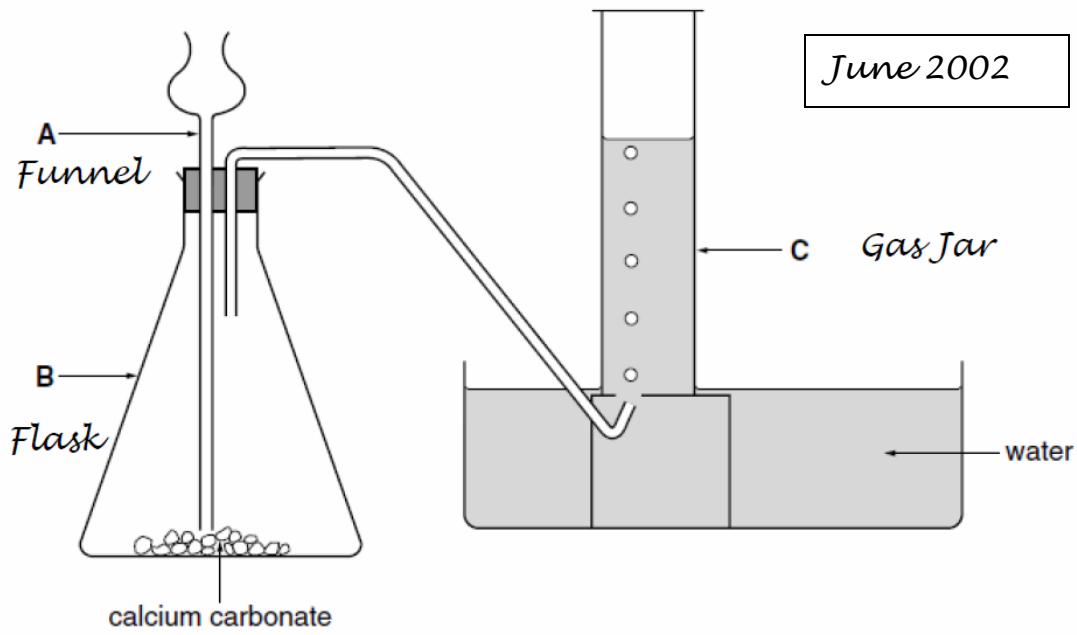


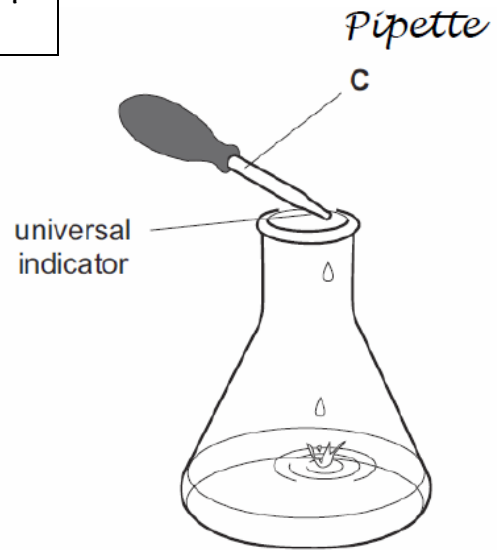
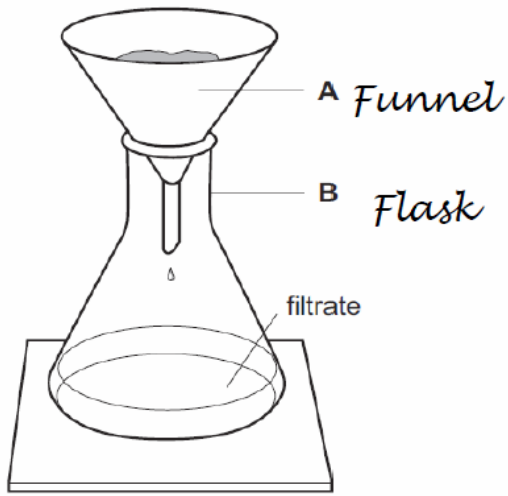
Nov2000/June2007/Nov2003



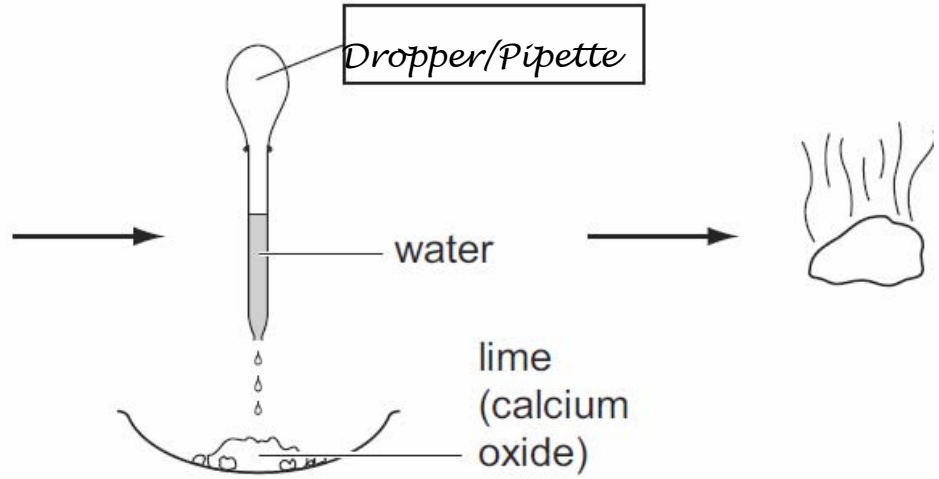
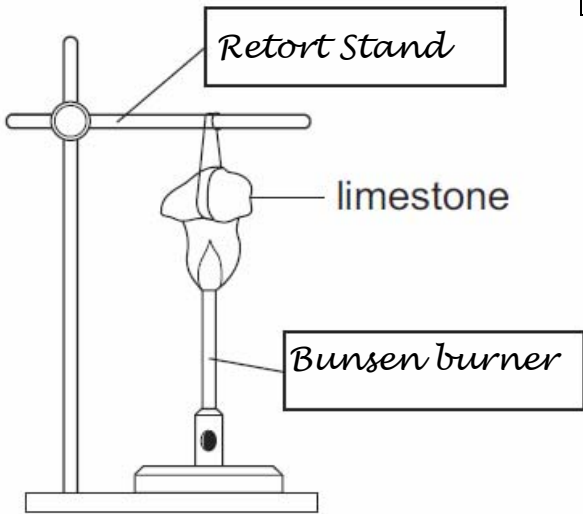
June 2002



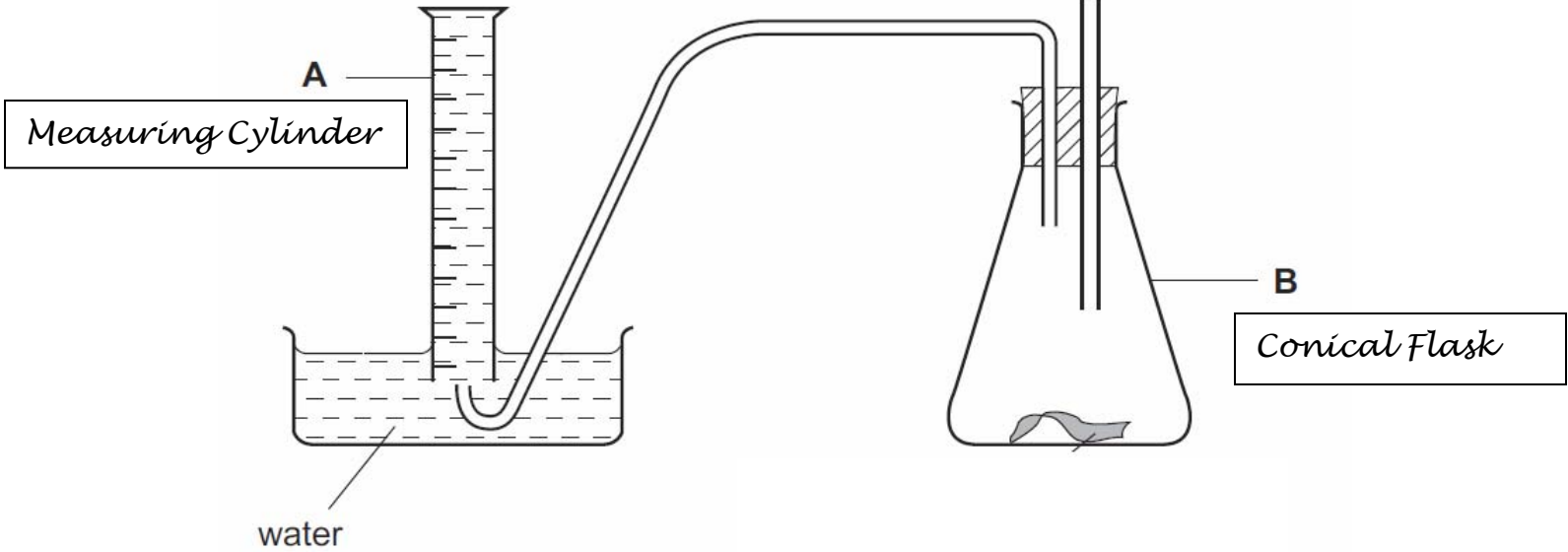
June 2004



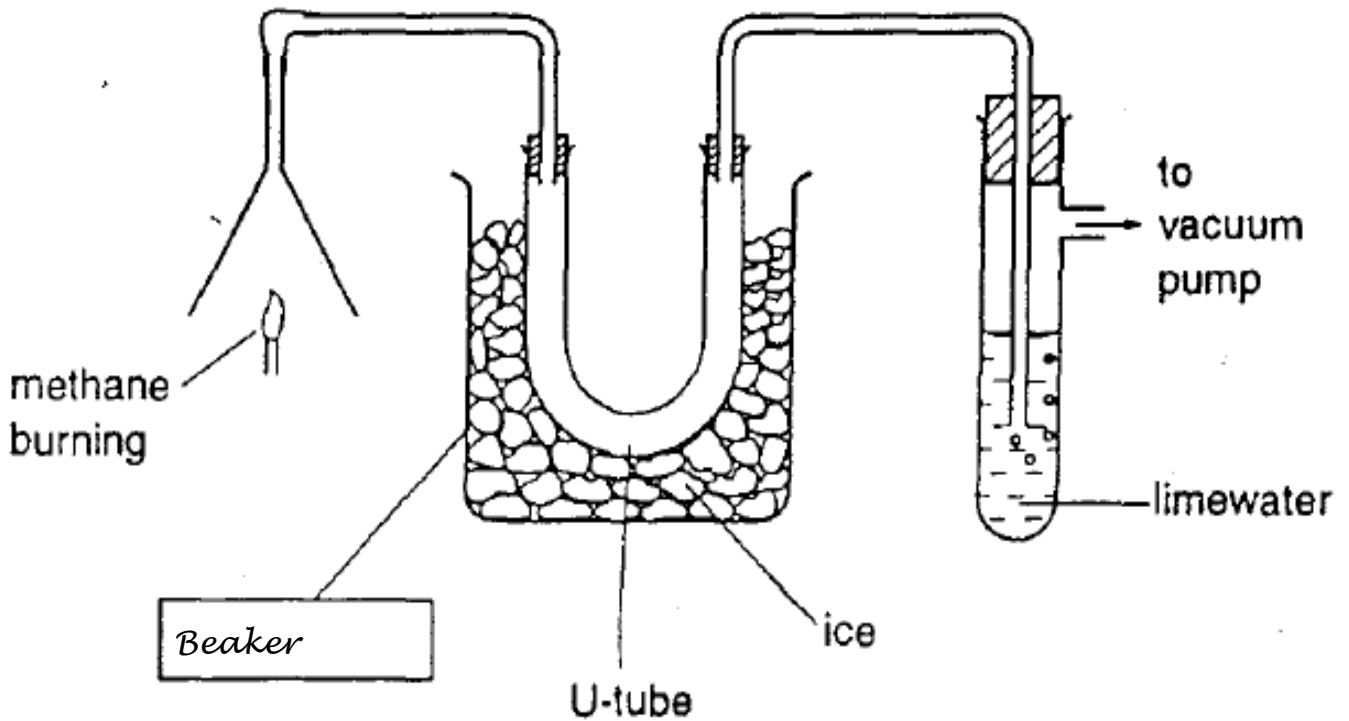
June 2005



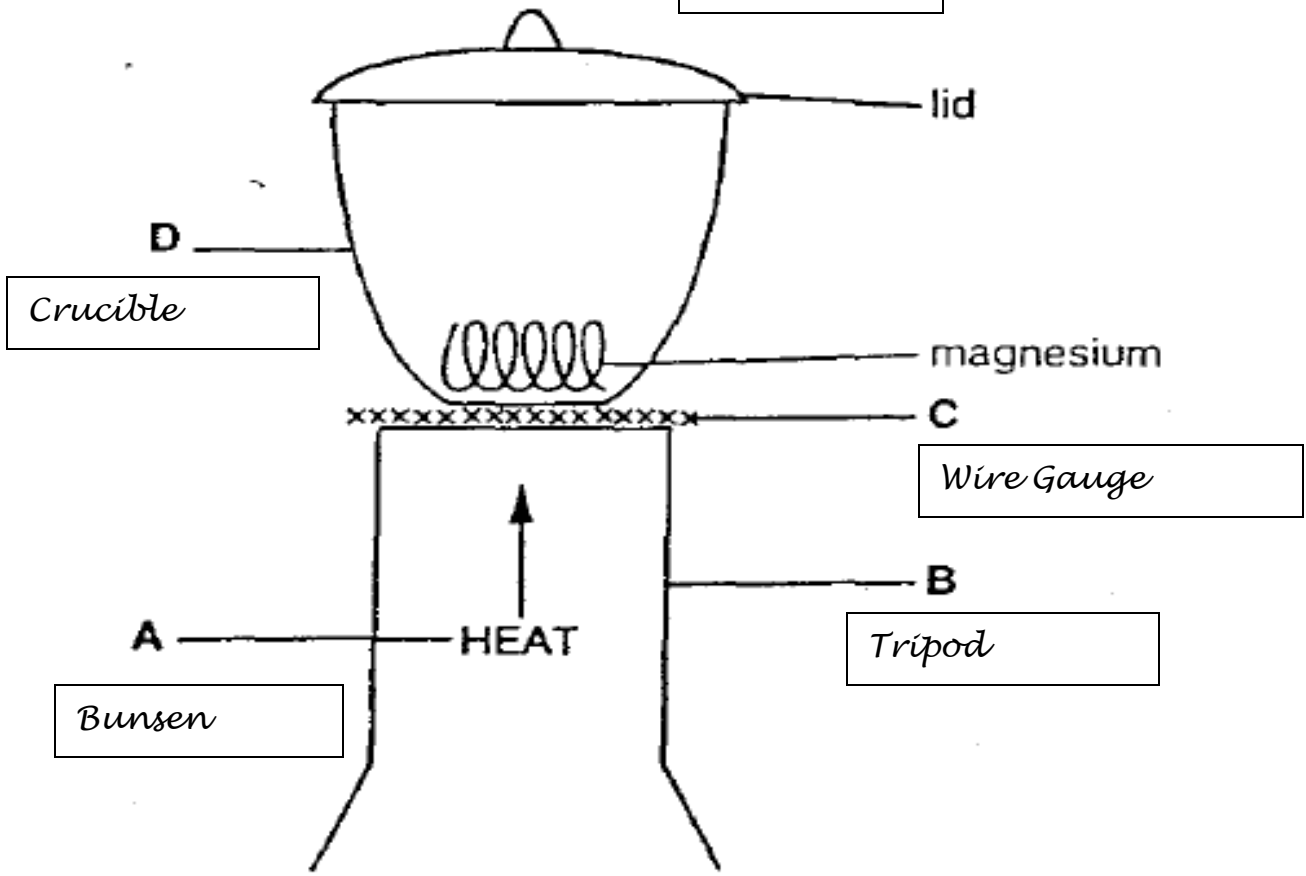
November 2004



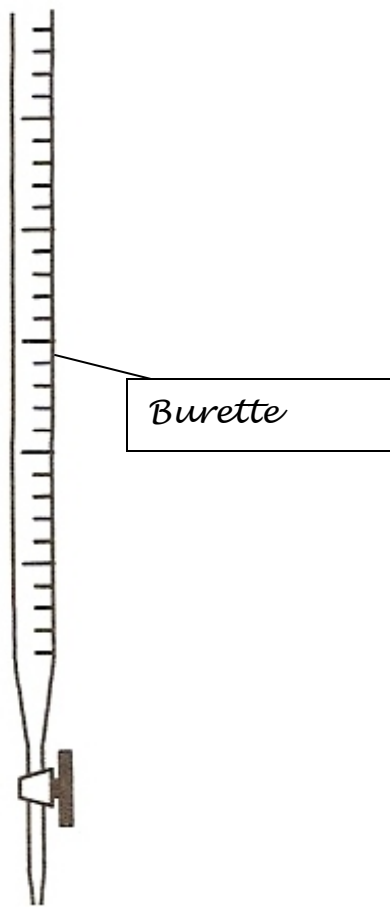
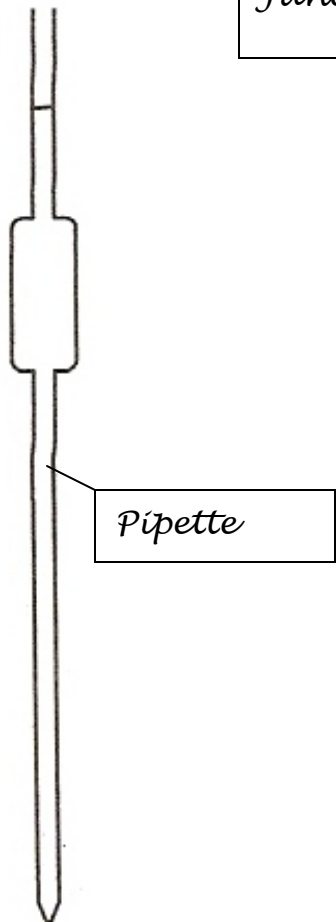
June 2001



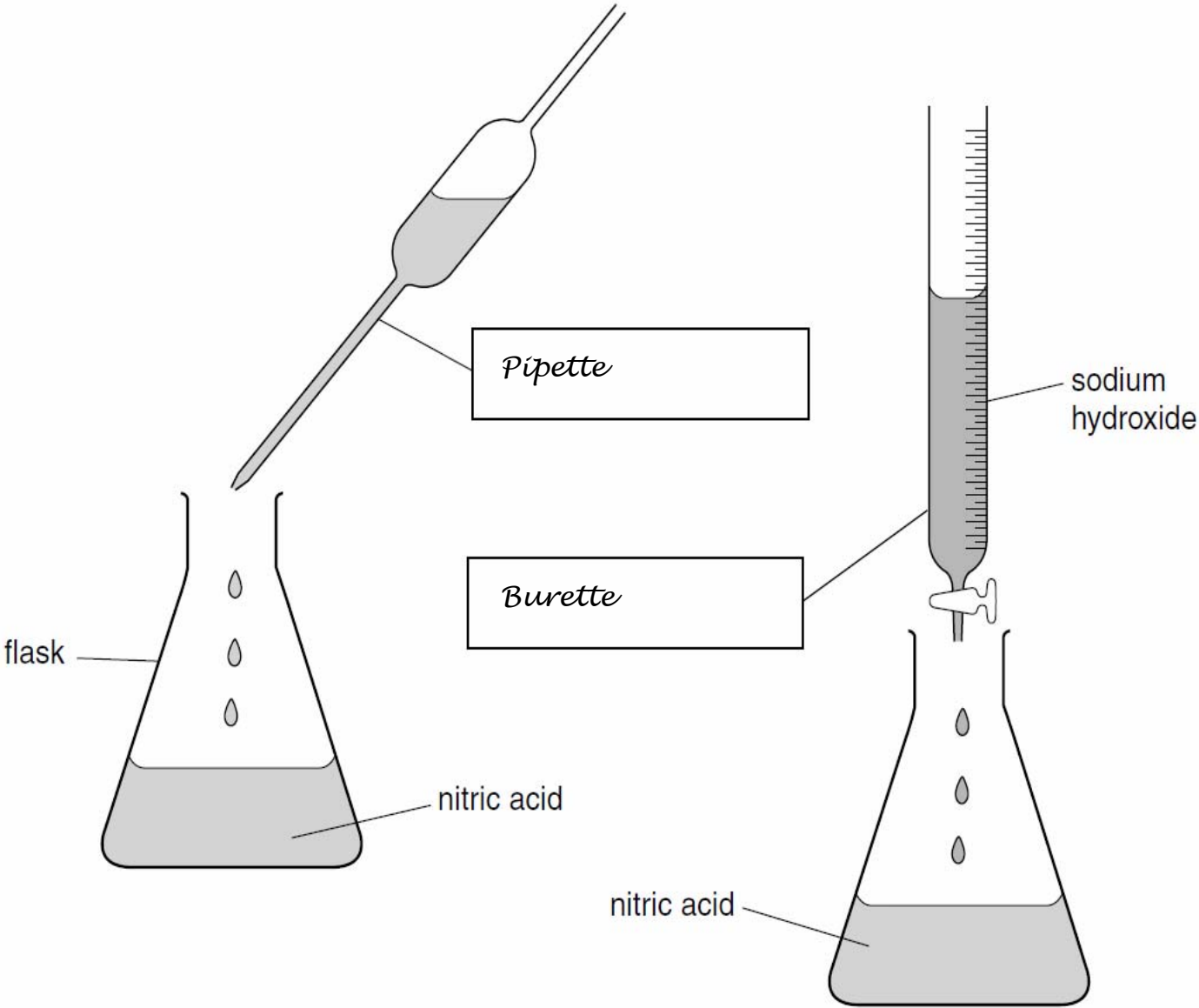
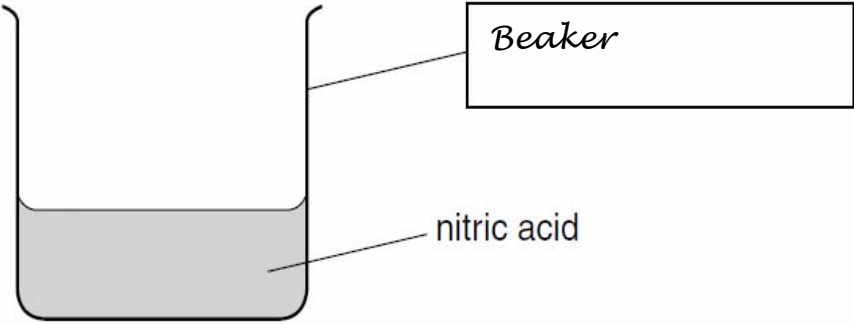
June 2000



June 1994

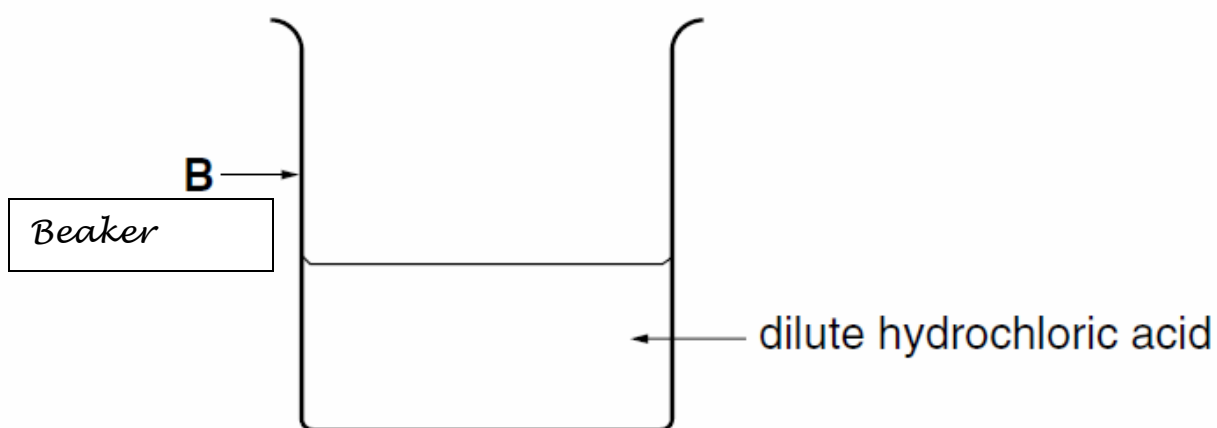
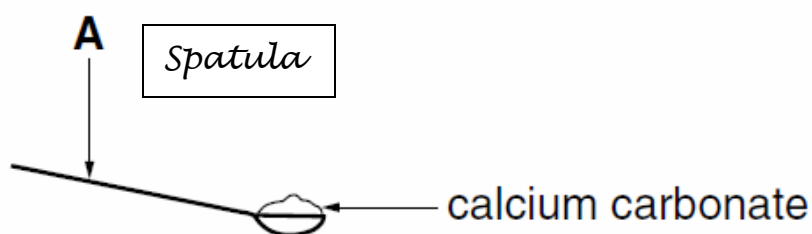


November 2001

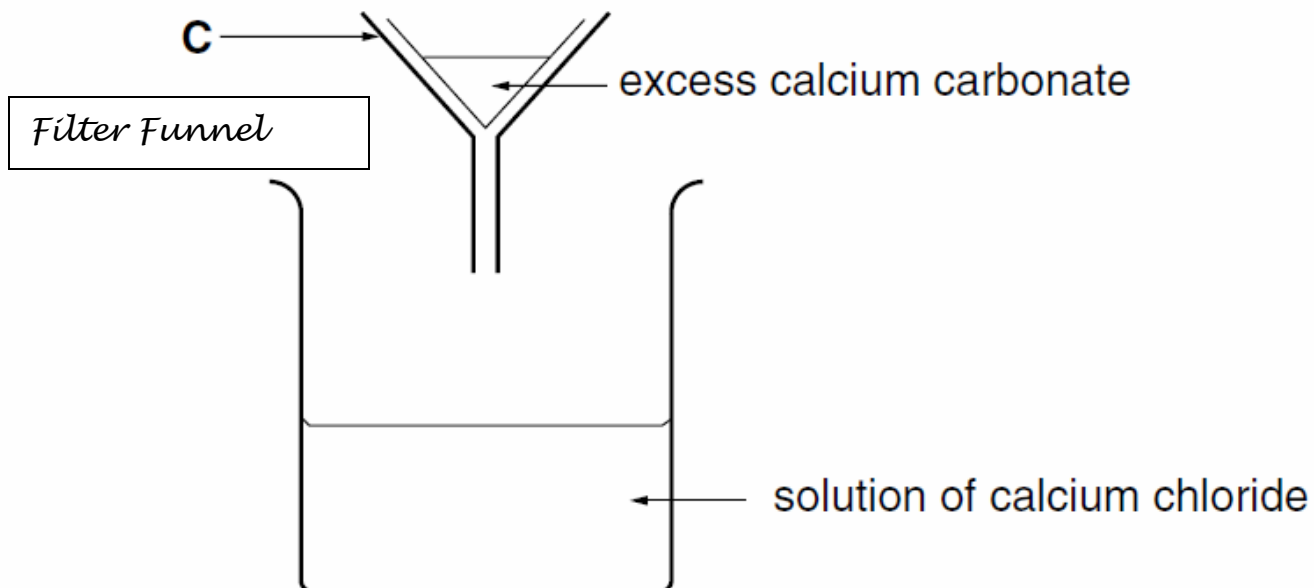


November 2002

Step 1 Excess calcium carbonate was added to hydrochloric acid.



Step 2. Excess calcium carbonate was removed from the solution.



Step 3. The solution of calcium chloride was tested with indicator paper.

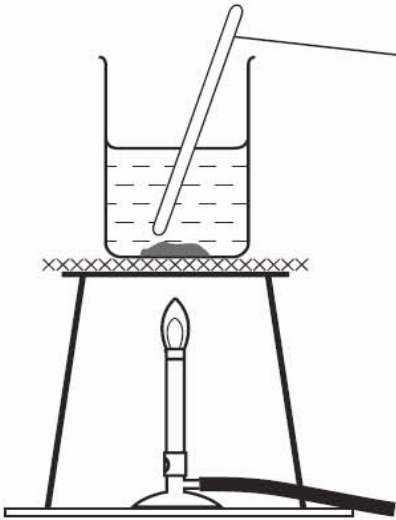
June 2003

A



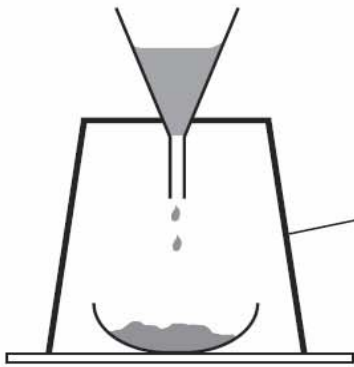
Mortar

B



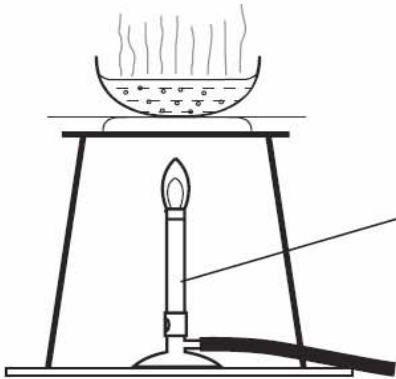
Stirrer/Stirring rod

C



Tripod

D

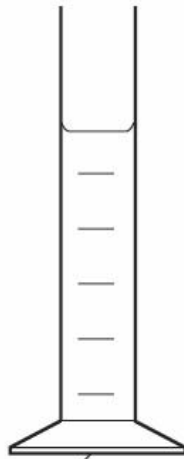
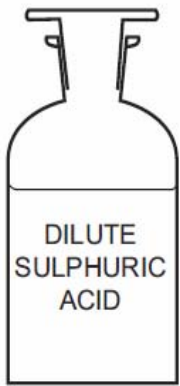


Bunsen burner

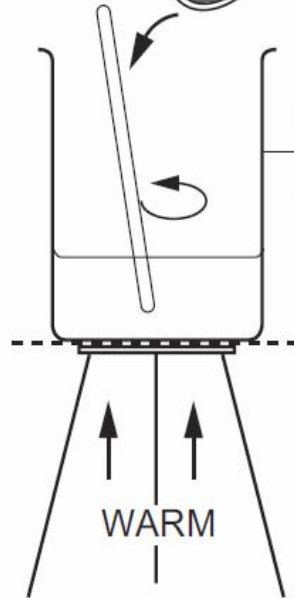
November 2005

Copper oxide was added until all the sulphuric acid had reacted

50 cm³ of dilute sulphuric acid was measured into a beaker



Spatula



Beaker

Measuring Cylinder

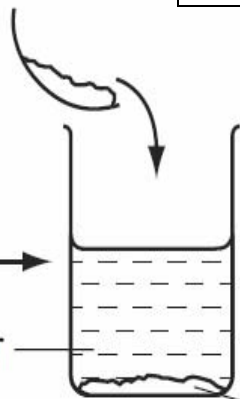
November 2006

Tongs



magnesium burning to form magnesium oxide

water



heat

magnesium oxide



Universal Indicator solution



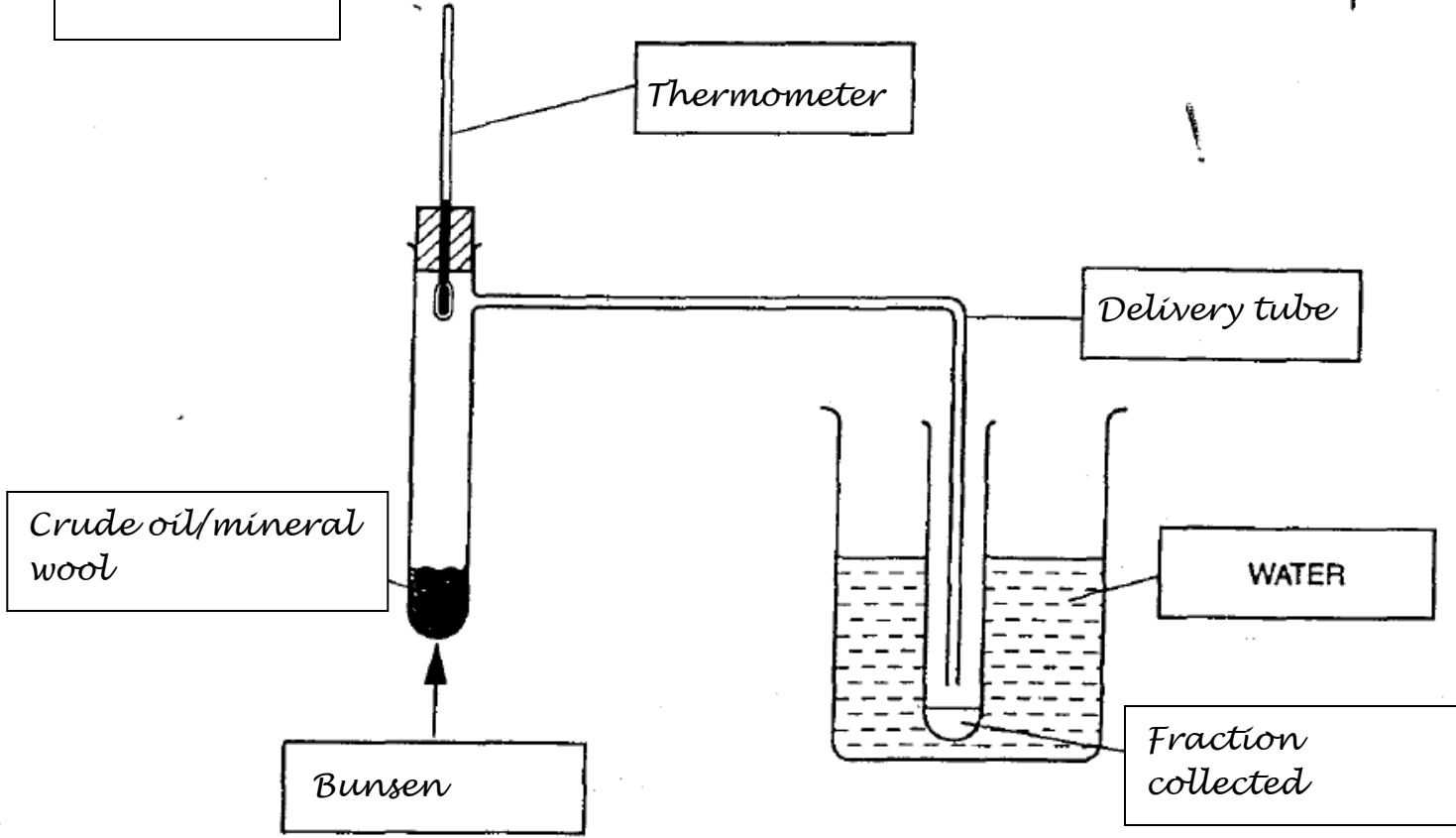
solution of magnesium hydroxide

indicator turns blue

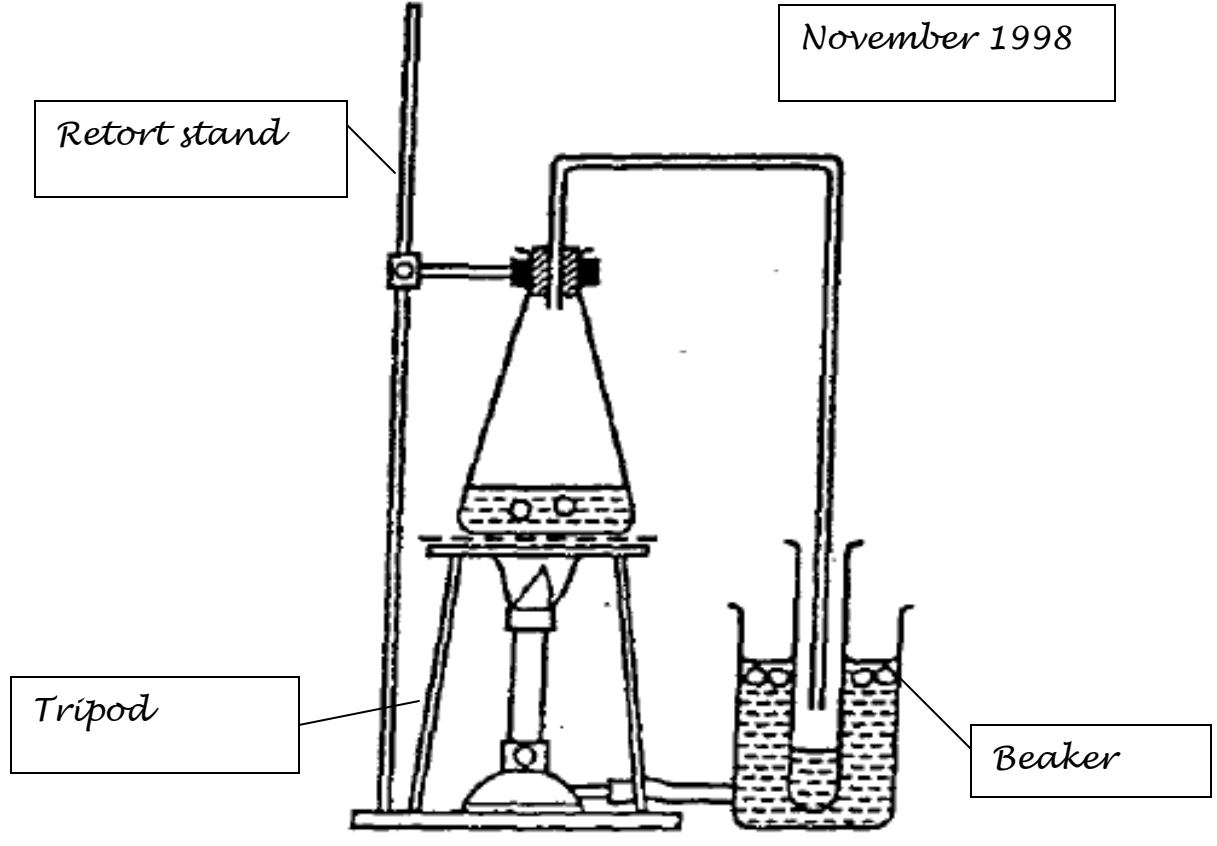
Beaker

Evaporating Dish

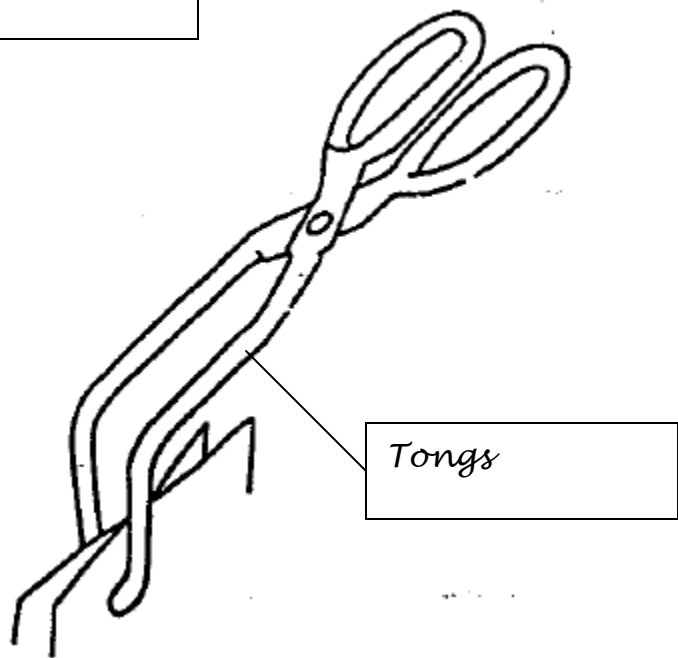
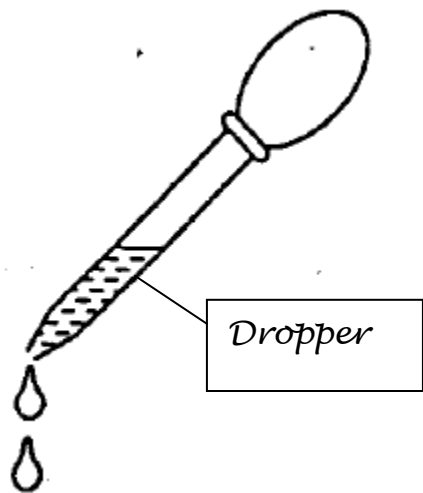
June 1999



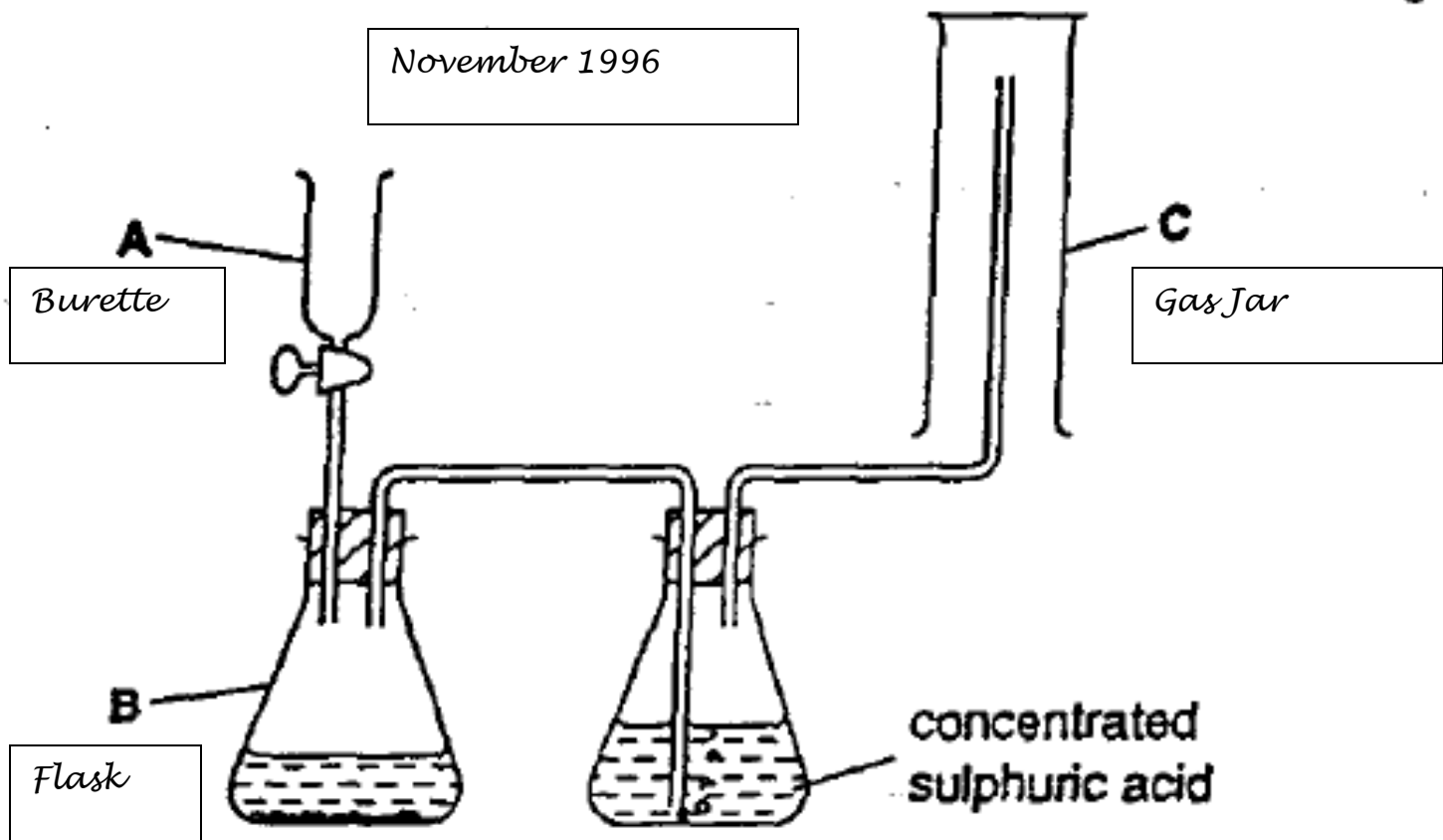
November 1998



November 1997

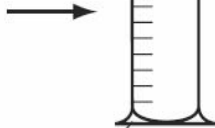
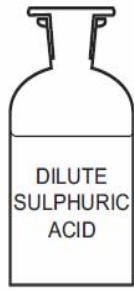


November 1996



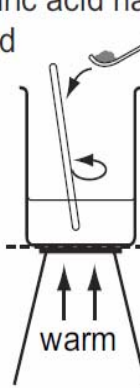
June 2008

50 cm³ of dilute sulphuric acid was measured into a beaker



Measuring Cylinder

copper oxide was added until all the sulphuric acid had reacted



Tripod Stand

Spatula



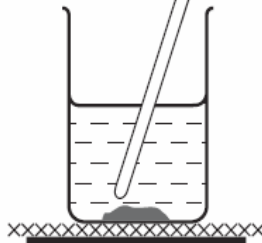
the mixture was filtered

November 2008

Pestle



Stirrer



Filter funnel

